

Notizen

Energies of Muonic Transitions in Cd and Br and Energies and Intensities of Nuclear Transitions in ^{78}Se and ^{80}Se

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The energies of the muonic 2-1 and 3-2 transitions in Cd and Br have been measured. The c -parameter of the (c, t) Fermi distribution was determined. Energies and intensities of nuclear gamma rays in Se are given.

From the measured muonic spectra of CdI_2 and CdBr_2 ¹ the energies of the muonic $2p_{3/2} - 1s_{1/2}$ and $3d_{3/2} - 2p_{1/2}$ transitions in Cd and Br could be determined. The energies were calibrated with the muonic transitions in iodine^{2,3} ($2p_{3/2} - 1s_{1/2}$: 3723.23 ± 0.20 keV; $3d_{3/2} - 2p_{1/2}$: 1150.42 ± 0.15 keV). A linear calibration was applied.

The energies of the muonic transitions in natural Cd and natural Br are:

$2p_{3/2} - 1s_{1/2}$ Cd	3268.8 ± 0.6 keV
$2p_{1/2} - 1s_{1/2}$ Cd	3229.1 ± 0.6 keV
$3d_{3/2} - 2p_{1/2}$ Cd	940.9 ± 0.3 keV
$3d_{5/2} - 2p_{3/2}$ Cd	905.9 ± 0.3 keV
$2p_{3/2} - 1s_{1/2}$ Br	2053.3 ± 0.4 keV
$2p_{1/2} - 1s_{1/2}$ Br	2040.6 ± 0.4 keV
$3d_{3/2} - 2p_{1/2}$ Br	494.6 ± 0.5 keV
$3d_{5/2} - 2p_{3/2}$ Br	482.4 ± 0.4 keV

Two nuclear gamma rays following the muonic capture in natural Br have been observed and assigned

(in parentheses intensities per muon captured in Br, corrected for the $1s_{1/2}$ life-time).

$$^{78}\text{Se}: 2^+ - 0^+ E_\gamma = 613.8 \pm 0.4 \text{ keV } (0.17 \pm 0.05),$$

$$^{80}\text{Se}: 2^+ - 0^+ E_\gamma = 666.3 \pm 0.4 \text{ keV } (0.16 \pm 0.05).$$

The gamma energies are in good agreement with previous measurements⁴⁻⁶.

So far, the nuclear charge distribution in Br^{nat} was not known. We can analyse our data in terms of a two-parameter (c, t) Fermi distribution for the nuclear charge density. Assuming $t = -2.3$ ⁷, we calculated numerically the energies of the $2p_{2/3} - 1s_{1/2}$ muonic transition in Br^{nat} for several values of c . To perform these calculations, we used a computer program⁸, and applied all necessary corrections as described in⁹.

The final value for c in Br^{nat} is

$$c = 4.7565 \pm 0.0035.$$

Proceeding analogously in the case of Cd^{nat} we obtained the value

$$c = 5.3804 \pm 0.0024.$$

However, we would like to point out that these values are of limited interest, since they describe only the average charge distribution of the natural isotopic mixture in the elements.

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